SUBJECT TEACHING GUIDE

G56 - Quantum Physics and the Structure of Matter II: Atoms, Molecules and Solids
Double Degree in Physics and Mathematics
Degree in Physics

Academic year 2018-2019

1. IDENTIFYING DATA					
Degree	Double Degree in Physics and Mathematics Degree in Physics			Type and Year	Compulsory. Year 3 Compulsory. Year 3
Faculty	Faculty of Sciences				
Discipline	Subject Area: Quantum Physics and the Structure of Matter Central Module				
Course unit title and code	G56 - Quantum Physics and the Structure of Matter II: Atoms, Molecules and Solids				
Number of ECTS credits allocated	6	Term		Semester based (1)	
Web					
Language of instruction	Spanish	English Friendly	No	Mode of delivery	Face-to-face

Department	DPTO. CIENCIAS DE LA TIERRA Y FISICA DE LA MATERIA CONDENSADA
Name of lecturer	IGNACIO HERNANDEZ CAMPO
E-mail	ignacio.hernandez@unican.es
Office	Facultad de Ciencias. Planta: + 3. DESPACHO PROFESORES (3010)
Other lecturers	JOSE ANTONIO ARAMBURU-ZABALA HIGUERA ANGEL MAÑANES PEREZ

3.1 LEARNING OUTCOMES

- Know the quantum description of hydrogen and helium atoms and learn how to generalize it to other atoms.
Understand the main atomic periodic properties and apply it to know the different the bonds of molecular and solid compounds.
Know the fundamentals of the photon absorption and emission by atoms.
Know how to find the symmetry operations and the symmetry group of a molecule or a solid.
Relate the symmetry of an atom, a molecule or a solid with its properties.
Understand the concept of irreducible representation and know how to use the Character Tables of molecular point groups.
To know the fundamental approximations for the resolution of the Schrödinger equation in simple molecules .
To understand the fundamentals and limitations of the adiabatic approach and its implications in the separation of electronic, vibrational and rotational states.
Know how to construct the molecular orbitals diagram of simple diatomic and polyatomic molecules, using symmetry.
Learn to deduce basic properties of the link.
Understand the microscopic differences between solid, liquid and gaseous states, state changes and phase transitions.
To know the concepts of Bravais network, motif, primitive cell, conventional cell, Wigner-Seitz cell and first Brillouin zone.
Know how to draw them in the case of simple structures.
To know how to analyze a Debye-Scherrer diagram of X-ray diffraction of a simple structure, studying the rules of selection and analyzing simply the intensity.
Know and apply the Born model for the ionic bond. Know how to apply the Born-Haber method.
Distinguish a solid insulator, semiconductor and metallic, relating the electrical conductivity to the location of the electrons and the scheme of electronic bands.
To know the basics of basic experimental techniques for the characterization of materials (solids, liquids and gases) and for the study of their electronic and vibrational structures: mass spectrometry, atomic absorption, infrared absorption, photoelectron spectroscopy, ...

4. OBJECTIVES

Obtain an overview of the microscopic constitution of matter, starting from the electronic structure of atoms, to understand why they interact to form molecules and solids.
Learn the quantum description of multielectronic atoms: The Helium atom
Understanding the independent particle model in multielectronic atoms and the need for antisimetrization of the wave function, the Hartree-Fock equations and the exchange energy concept.
Describe the states of a multielectronic atom when including spin-orbit and hyperfine interactions.
Understand the atom-radiation interaction.
Know how to calculate transition probabilities in electric dipolar approximation.
To understand from the first principles the microscopic origin of the molecular bond and the intermolecular forces , understanding the similarities and differences between the ionic and covalent bonds , as well as the intermolecular interactions of van der Walls and the hydrogen bonds .
Be able to analyze and understand the microscopic origin of many relevant macroscopic properties (molecular reactivity, electrical conductivity, thermal expansion, piezoelectricity, hardness, ...).
Acquire and understand the basic knowledge of the structure of crystalline solids, their symmetries and their structures of electronic bands.
Understand the concepts and foundations of theories and models of the structure of matter, their approximations and limits, as well as the orders of magnitude of the properties analyzed.
Acquire the fundamentals of relevant materials characterization techniques (solids, liquids and gases), paying special attention to X-ray diffraction in crystalline solids.

6. COURSE ORGANIZATION

CONTENTS	
1	One electron atoms
2	Atoms of many electrons
3	Emision and absorption of photons by atoms
4	Molecule: Atomic bonds
5	Electronic molecular states
6	Molecular symmetry
7	Rotational and vibrational molecular spectra
8	Intermolecular forces
9	Crystalline structure of solids
10	Reciprocal laticce and Xray diffraction
11	Partial evaluation .Chapters 1-5
12	Partial evaluation .Chapters 6-10
13	Final evaluation in February

7. ASSESSMENT METHODS AND CRITERIA

Description	Type	Final Eval.	Reassessn	%
Topics 1-5: Theoretical questions and practical problems	Written exam	No	Yes	50,00
Topics 6-10: Theoretical questions and practical problems	Written exam	No	Yes	50,00
February: Theoretical questions and practical problems	Written exam	Yes	Yes	0,00
September: Theoretical questions and practical problems	Written exam	Yes	No	0,00
TOTAL				100,00

Observations

Observations for part-time students

The professors will take care of special cases of studens at partial time dedication (Working or high level sportives, etc)

8. BIBLIOGRAPHY AND TEACHING MATERIALS

BASIC

- R. Eisberg y R. Resnick, Física Cuántica. Átomos, Moléculas, Sólidos, Núcleos y Partículas, (Limusa, Noriega Editores, México, 2000).
- J. J. Brehm, and W. J. Mullin, Introduction to the Structure of Matter. A course in Modern Physics (Wiley, New York, 1989)
- P. W. Atkins, J. de Paula, Physical Chemistry (Oxford U.P., 9th ed., 2010)